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21: Acyl Transfer~~
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CHEM-002A Ch 21

Enolate Problems

102 Ch 21 Enzymes

Organic Chemistry II -

Chapter 21 - Acyl

Transfer

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Organic Chemistry

(Part 2)~~Chapter 21:~~

~~Organic Chemistry~~

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Chapter 21 Part A

~~CHEM 2320 Chapter~~

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~~Chapter 21-~~

~~Hydrocarbons~~

~~Chapter 21 Amine~~

~~Chemistry: Part 1 of 4~~

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~~Zoom Functional~~

~~Groups Memorizing~~

~~Tricks Acidity of~~

~~Alcohols Reactions of~~

~~Amines Do not be~~

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Determining EDG or
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Structure and
Bonding Chapter 21
—Amine Chemistry:
Part 1 of 2~~

Chapter 21 Part 1 -
Acidity of Phenols

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Chapter 21: Organic
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CHEM-002A

~~Frankenstein~~

~~Chapter 21~~ Chapter
21 – Amine

Chemistry: Part 2 of 4

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~~Chemistry Chapter 21~~

Organic Chemistry,

7e (Wade) Chapter

21 Carboxylic Acid

Derivatives 1) N-

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Methylacetamide is an example of: A) a primary amide B) a secondary amide C) a tertiary amide D) an N, N-disubstituted amide E) an imine

Answer: B Diff: 1

Section: 21.2 2)

Cyclic amides are called: 3) Amides are less basic than amines because: 4)

Cyclic esters are

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called: 1

Chemistry

~~chapter 21 - Organic~~

~~Chemistry 7e (Wade~~

~~Chapter 21 ...~~

The full step-by-step
solution to problem:

PROBLEM 21-13

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answered by , our top
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Carboxylic Acid
Derivatives 21.1
Introduction 21.2
Structure and

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Nomenclature of Acid
Derivatives 21.3

Physical Properties of
Carboxylic Acid
Derivatives 21.4

Spectroscopy of

Carboxylic Acid
Derivatives 21.5

Interconversion of

Acid Derivatives by
Nucleophilic Acyl
Substitution 21.6

Transesterification

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... 21 Chapter 17 21

Alkyl Group

Stabilization CH_2CH_3

Br_2 FeBr_3 CH_2CH_3

Br CH_2CH_3 Br

CH_2CH_3 $\text{Br} + +$ o-

bromo (38%) m-

bromo (< 1%) p-

bromo (62%) Alkyl

groups are activating

substituents and

ortho, para-directors.

This effect is called

the inductive effect ...

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Organic

~~17 - Reactions of
Aromatic Compounds~~

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Organic molecules must have N-H or O-H to be able to form a hydrogen bond. iii. The bond is polar and the H becomes positively charged (Electrophile- likes electrons). iv.

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stry Chapter 2 Outli
ne Acids and Bases
Chapter 21~~

Mohrig, J.R.,
Hammond, Schatz,
P.F. and Morrill, T.C.,
Modern Projects and
Experiments in
Organic Chemistry,
Freeman, New York,
2003. Course
Description and
Objectives: Chemistry

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202 is a course that will build on the foundation that was established in CHM 201. The course will begin with the study of alcohols, their synthesis and reactions.

~~Bill Price – Chemistry
202 → La Salle
University~~

1.1: The Origins of

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Organic Chemistry

The term "organic chemistry" was first used in the 1800's to distinguish reactions performed by

chemists from those occurring within living organisms. 1.2:

Principles of Atomic Structure (Review)

Atomic structure is applied to several topics of organic

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~~1: Introduction and
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Chapter 13: NMR 113

Terms.

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PLUS. Chemistry 14C

Definitions Midterm 1

(Lecture 9 - 17) 40

Terms. ... Wade

Chapter 12 43 Terms.

tiffunnyw PLUS.

Enthalpy 6 Terms.

Jennie_Scott. Gibbs

Free Energy 11

Terms.

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NMR 19 Terms.
haleighmiller366
PLUS. Features.
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Infrared Spectroscopy
and Mass
Spectrometry 2.

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Chapter 12 2

Introduction •

Spectroscopy is a technique used to determine the structure of a compound.

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By Leroy G. Wade -
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Students often say,
“ I studied 40 hours
for this exam and I
still didn ’ t do well.
Where did I go
wrong? ” Most
instructors hear this
complaint every year.
In many cases, it is
true that the student
invested countless

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hours, only to produce abysmal results. Often, inefficient study habits are to blame. The important question is: why do so many students have difficulty preparing themselves for organic chemistry exams? There are certainly several factors at play here,

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but perhaps the most dominant factor is a fundamental disconnect between what students learn and the tasks expected of them. To address the disconnect in organic chemistry instruction, David Klein has developed a textbook that utilizes a skills-based approach to

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instruction. The textbook includes all of the concepts typically covered in an organic chemistry textbook, but special emphasis is placed on skills development to support these concepts. This emphasis upon skills development will provide students with a greater opportunity

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to develop proficiency in the key skills necessary to succeed in organic chemistry. As an example, resonance structures are used repeatedly throughout the course, and students must become masters of resonance structures early in the course. Therefore, a

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A significant portion of chapter 1 is devoted to drawing resonance structures. Two chapters (6 and 12) are devoted almost entirely to skill development. Chapter 6 emphasizes skills that are necessary for drawing mechanisms, while chapter 12 prepares the student for proposing

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syntheses. In addition, each chapter contains numerous

Skillbuilders, each of which is designed to foster a specific skill.

Each

skillbuilder contains three parts: 1. Learn the Skill: a solved problem that demonstrates a

particular skill; 2. Practice the Skill:

Practice the Skill:

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numerous problems (similar to the solved problem) that give the students an opportunity to practice and master the skill; 3. Apply the Skill: one or two more-challenging problems in which the student must apply the skill in a slightly different environment. These problems include

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Conceptual, cumulative, and applied problems that encourage students to think out of the box. Sometimes problems that foreshadow concepts introduced in later chapters are also included. All SkillBuilders are visually summarized at the end of each chapter (Skillbuilder

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review), followed by a list of suggested in-chapter and end-of-chapter practice problems. This text is an unbound, three hole punched version.

Acclaimed for its clarity and precision, Wade's Organic Chemistry maintains scientific rigor while engaging students at

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Critically and avoid
taking "short cuts" to
solve problems.

Mechanism Boxes
and Key Mechanism
Boxes strengthen
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understanding of
Organic Chemistry as
a whole while
contemporary
applications reinforce
the relevance of this
science to the real

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inorganic and

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