

Chemistry Chapter 12

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Chemistry Chapter 12

1. increasing the surface area. 2. increasing the temperature. 3. increasing the strength of the intermolecular forces. 1 and 2 only. Liquids that have high vapor pressure and low boiling points are called. volatile liquids. The amount of heat required to melt one mole of a solid is called the: heat of fusion.

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Answer Key Chapter 12 - Chemistry 2e | OpenStax 1. The instantaneous rate is the rate of a reaction at any particular point in time, a period of time that is so short that the concentrations of reactants and products change by a negligible amount. The initial rate is the instantaneous rate of reaction as it starts (as product just begins to form).

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Class 12 Chemistry Chapter 7 P – Block Elements Notes- Pdf Download P-Block elements are seen on the right side of the periodic table and consist of six columns. They are so called because in these the valence electron which is commonly known as outermost electron enters the p-orbital.

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