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1. \_\_\_\_\_

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Entropy, and Free  
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Hunger Games

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And

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Test Practice. Your

Results: The correct answer for each

question is indicated by a . 1:

Hyperventilation can cause ...

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PLAY. Properties of  
acids. sour taste feel  
watery turn litmus  
paper red corrosive

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strong or weak  
electrolytes react  
with ...

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454 CHAPTER 15

Formula Acid name

HF hydrofluoric acid

HCl hydrochloric acid

HBr hydrobromic acid

HI hydriodic acid H<sub>2</sub>S

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hydro-sulfuric acid

TABLE 15-1 Names of Binary Acids  
FIGURE

15-2 A strip of pH paper dipped into vinegar turns red, showing that vinegar is an acid.

---

CHAPTER 15 Acids and Bases

Chapter 15 Acids and Bases. STUDY. PLAY.

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Chapter 15 main  
idea. acids and bases  
change the color of  
compound  
indicators. Acids.  
were first recognized  
as a ...

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Test. PLAY. Match.

Gravity. Created by.

ralexis16. Terms in  
this set (43) All of the  
following lipids can  
be hydrolyzed to  
form fatty acids  
except ? Steroids.

What is a saturated  
fatty acid? They do

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## Chapter 15 Test

not contain a C=C double bond in the carbon chain. What characterizes a triacylglycerol?

---

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Chemistry Concepts  
and Applications

Chapter 15: Acids and  
Bases React

Interactive Tutor.

True and False Game

---

Interactive Tutor

Chapter Fifteen: Acid-  
Base Titration and pH

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If you look around the room while titrations are happening, there are people standing like this: and the best part is that you probably will too, because it works.

---

Chapter Fifteen [Acid-Base Titration and pH]

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This chapter 15 test acids and bases answers, as one of the most working sellers here will categorically be in the middle of the best options to review.

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Chapter 14/15

*Page 24/34*

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## Chapter 15 Test

Equilibrium and Acid-Base Practice Test.

Multiple Choice

(Choose the best

answer.--Answers

Found At End of Test)

\_\_\_\_\_ 1. The

reaction:  $3 \text{CuCl}_2 (\text{aq})$

$+ 2 \text{Al} (\text{s}) \rightarrow 3 \text{Cu} (\text{s}) + 2$

$\text{AlCl}_3 (\text{aq})$  has the

following  $K_c$

expression: a)  $[\text{CuCl}_2]^3$

$[\text{Al}]^2 / \{ [\text{Cu}]^3$

$[\text{AlCl}_3]^2 \}$  b)  $[\text{Cu}]^3$

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## Chapter 15 Test

[AlCl<sub>3</sub>]<sub>2</sub> / { [CuCl<sub>2</sub>]<sub>3</sub> [Al]<sub>2</sub> } c) [CuCl<sub>2</sub>]<sub>3</sub> / [Cu]<sub>3</sub> d) [Al]<sub>2</sub> / [AlCl<sub>3</sub>]<sub>2</sub> e) None of these

---

Chapter 14/15

Test--Equilibrium and  
Acids/Bases

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memorize the terms,

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Ch. 14-15 Acid & Base  
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1. Aqueous solutions of acids have a sour taste. 2. Acids change the color of acid-base

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## Chapter 15 Test

Acids and Bases Answers

indicators. 3. Some acids react w/active metals to release hydrogen gas, H<sub>2</sub>. 4. Acids react with bases to produce salts and water. (salt- the reaction products are water and an ionic compound called a salt). 5. Some acids conduct electric current. 6 ...

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### Acids And

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Answers to Chapter  
15 & 16 Study  
Questions. 1. strong  
acid + strong base:  
steepest at  
equivalence, low pH  
at start depending on  
concentration of acid,  
equivalence at pH 7,

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## Chapter 15 Test

no buffering. weak

acid + strong base:

less steep at

equivalence than

strong acid, low pH at

start depending on

and concentration of

acid, equivalence at

basic pH, buffering.

weak base + strong

acid: less steep at

equivalence than

strong base, high pH

at start depending on

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## Chapter 15 Test

and concentration of  
base, equivalence at  
acid pH, ...

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controlled addition

and measurement of

the amount of a

solution of known

concentration

required to react

completely with a

measured amount of

a solution of

unknown

concentration.

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